

Below is a summarized overview regarding the **chemical resistance** of Radici Group High Performance Polymers products, categorized by polymer types. Each **column** represents a different **base polymer**, while each **row** lists a common **chemical agent** in its typical form and concentration that can come into contact with the polymer. The resistance against each chemical agent is assessed for the base polymer alone and, unless otherwise specified, at room temperature and for short-term exposure (up to some hours).

Chemical resistance is qualitatively evaluated by rating each polymer/chemical agent combination as follows:

● ● = Excellent

The material is not attacked by the medium which causes no irreversible damage to the polymer. There may be little change in weight, small effect on dimensional or mechanical properties.

● = Good

The exposure to the medium may cause a slight change in properties and/or surface aesthetics of the polymer, but in general it remains suitable for most practical uses.

= Fair/Limited

Limited resistance; according to actual exposure conditions there may be noticeable changes in properties and/or surface aesthetics; prolonged exposure to the medium can lead to irreversible damage.

x = Not resistant/Soluble

The chemical agent attacks the polymer, causing degradation, dissolution and/or major damaging phenomena such as stress cracking; typically, irreversible damage can occur even with a short time exposure.

Chemical resistances here reported must be considered as a mere preliminary indication and used only for screening purposes.



Polymers behaviour

Generally, compounds (reinforced, heat stabilized, impact modified or otherwise formulated grades) based on the same polymer matrix will exhibit similar behaviour. However, depending on the specific case, **differences are also possible**, even to a significant extent.

The actual behaviour of a material in contact with a chemical agent can vary considerably **depending on multiple factors** such as temperature, duration of exposure, concentration of the medium, surface roughness and integrity, contacting parts shape and design, etc.

Additionally, the suitability for a specific purpose of a component made in polymeric material may or may not be significantly affected by **secondary effects of chemical exposure**, such as swelling, absorption, modification or loss of weight and mechanical properties, discoloration, etc. according to the nature and function of the component itself.

For all these reasons, it is always strongly recommended to **conduct specific tests** on the actual parts in contact with the medium of interest under real working conditions, or in conditions as close as possible to the expectable ones.







		1	radilon* A torzen*		on° S	radilo)n ° D	radilon [,] DT		radilon* Aestus				raditeck* P	
		PA6	PA66		PA6		PA610		12	PF	PΑ	PI	ВТ	PP	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
ACIDS (INORGANIC)									'				'		'
	Diluted (<10%)	•	01	•	01	•		•		•		•••		•••	
Sulfuric Acid	Concentrated (<38%)	x		x		X		x		x		• •		• •	
	Concentrated	x		x		X		x		x		x		• •	
Nitric Acid	Diluted (<10%)	x		x		x		x		•		•••		•••	
THE AGE	Concentrated	х		х		x		х		х		x		•	
Boric Acid	10%	•		•		•		••		•••		•••			
Chromic Acid	Diluted (<1%)	•		•		•		•		• •		•••		•••	
Official Acid	Concentrated (<40%)	x		х		x						•		•	
Hydrochloric Acid	Diluted (<10%)	x		x		X		•		•		•••		•••	
Tryatochione Acid	Concentrated	x		х		x		x		x		x			
Phosphoric Acid	10%	x		х		x		х		•		х		•••	





		radilo torz		radilo	on°s	radilo	on ° D	radi lo	n° DT	radilor	1 ° Aestus	radit	er [®] B	radite	ck* P
		PA6	56	PA6		PA6	10	PA6	512	PF	PA	PE	T	PP	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
ACIDS (ORGANIC)											'				
	5%	••	02	• •	02	• •		••		• •		•••		•••	
Acetic Acid	10%	•		•		• •		• •		• •		•••		•••	
	Concentrated	x		x		X		x		•		•		•••	
Citric Acid	10%	••	03	••	03	••		•••				•••		•••	
Gittle Acid	Concentrated (>50%)	•	04	•	04	x									
Formic Acid	Diluted	•	01	•	01	••		••		••		•••		•••	
Formic Acid	Concentrated (>10%)	x		x		x		x		x		•		• •	
Benzoic Acid (aqueous sol)	Saturated	x		х		•		••		••		•••			
Acrylic Acid	Concentrated	x		x		x									
Oleic Acid		•••		•••		•••		•••		•••		•••		•••	
Lactic Acid	Diluted (<10%)	•••		•••		•••		•••		• •		•••		•••	
Lactic ACIU	Concentrated (>90%)	х		X		•		•		• •		X		•••	
Stearic Acid		•••		•••		•••		•••		•••		•••			
Uric Acid	20%	•••		•••		•••				•••		•••			





		l	radilon [•] A torzen [•]		radilon s		radilon [*] D		radilon [*] DT		* Aestus			radite	eck* P
		PA6	PA66		PA6		PA610		12	PP	Ά	PE	ВТ	PP	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
BASES															
Ammonia	10%	•••		•••		•••		•••		•••		•••		•••	
	1%	•••				•••						•		•••	
Sodium Hydroxide	10%	•••	05	•••	05	•••		•••		•••		•		•••	
	50%	•		•		•••		•••		•••		x		•••	
Potassium Hydroxide	10%	•••		•••		•••		•••		•••		•		•••	
Fotassium nyuroxide	50%	•		•		••		••		•••		x		•••	
Ammonium Hydroxide	10%	•		•						••		•		•••	
Calcium Hydroxide	Saturated	•••		•••		•••						•••			
Lithium Hydroxide	10%	•••		•••		•••						•••			
Chloramines	<10%	•	06	•	06	•	06	•	06	•••		•			





			radilon [•] A torzen [•]		on°s	radilo	on ° D	radil o	n° DT	radi lor	1 ° Aestus	raditer в	radite	ck* P
		PA6	6	PA6		PA6	10	PA6	12	PF	PA	PBT	PPS	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank Notes	Rank	Notes
INORGANIC SUBSTANCES												· ·		
7: 0.1:-1	10%	•	07	• •	07	•••		•••		•••		•••	•••	
Zinc Chloride	>30%	x		•		• •		• •		•••			•••	
Magnesium Chloride	10%	•••	08	•••	08	•••		• •				•••	•••	
Calcium Chloride	10%	••	07	• •	07	•••		•••		•••		•••	•••	
Calcium Chloride (alcoholic sol.)	20%	x		x		•		•		•				
Potassium Bromide	10%	• •	09	• •	09	• •						•••		
Ondiana Oblanida	10%	•••	08	•••	08			•••		•••		•••	•••	
Sodium Chloride	Saturated	• •	07	• •	07	• • •		•••		• • •			•••	
Sodium Carbonate	Diluted (<10%)	•••		•••		•••		•••	17			•••	•••	
Sodium Hypoclorite	10%	•	01	•	01	•	01	•	01	•		• •	•••	
Oxygen (gas, low pressure)		•••		•••		•••		•••		•••		•••		
Ozone (gas)	<5ppm	•••		•••		•••		•••		•••		•••		
	1%	• •	06	• •	06					•••		•••	•••	
Hydrogen Peroxide	3%	•	06	•	06	•	06	• •	06	• •		•••		
	30%	х		х		х		х		х		•••	• •	





		l	radilon [*] A torzen [*]		radilon [*] S		radilon [,] D		radilon [,] DT		1 ° Aestus	radit	er [*] B	radite	eck* P
		PA6	PA66		PA6		10	PA6	12	PF	PA	PE	ВТ	PP	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
HYDROCARBONS															
Cyclohexane		•••		•••		•••		•••		•••		•••		•••	
Benzene		•••		•••		•••		•••		•••		•••			
Toluene		•••		•••		•••		•••		•••		•••		• •	
Methane (gas)		•••		•••		•••		•••		•••		•••		•••	
Iso-octane		•••		•••		•••		•••		•••		•••			
n-Hexane		•••		•••		•••		•••		•••		•••		•••	
Xylene		•••		•••		•••		•••		•••		•••		•••	
FUELS															
Diesel		•••		•••		•••		•••		•••		•••		•••	
Bio-Diesel		•••		•••		•••		•••		•••					
Gasoline		•••		•••		•••		•••		•••		•••		•••	
Gasoline with EtOH	10-15%	••	10												
Kerosene		•••		•••		•••		•••		•••		•••		•••	





		radilo torz		radilo	on° S	radilo)n ° D	radi lo	n ° DT	radi lor	l* Aestus	radi t	er [®] B	radite	eck* P
		PA6	PA66		PA6		10	PA6	12	PF	PΑ	PB	Т	PP	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
AUTOMOTIVE FLUIDS															
Transmission Oil		•••				•••				•••				•••	
Aqueous Urea Solution	Commercial %	•••	11			• •		• •		• • •		•••		•••	
Cooling Liquids based on Glycols	50%	•	12	•	12	• •		• •		•••				•••	
Lubricants - mineral oils		•••		•••		•••		•••		•••		•••		•••	
Brake Fluid		•••		•••		•••		•••		•••		•••		•••	
ALCOHOLS															
Ethanol	Diluted	•••						•••							
Ethanoi	Concentrated	• •	13	• •	13			• •		• • •		• • •		• • •	
Methanol	Concentrated	• •	13	• •	13	• •		• •		• •		•••		•••	
Benzyl Alcohol		•		•		•		•		•					
Phenol	Concentrated	х		х		х		х		•		х		•••	
Butanol	Concentrated	•••		•••		•••		•••		•••		•••		•••	
Cyclohexanol		• •		• •		•••		•••				•••		•••	
Glycerol		• •	14	• •	14	• •	14					•••			
Isopropanol		• •		• •		• •		• •		•••		•••			





			radilon [•] A torzen [•]		radilon [*] S		radilon [,] D		radilon [,] DT		1 ° Aestus	radit	er* B	radite	ck* P
		PA6	6	PA	PA6		PA610		12	PF	PA	PE	ВТ	PP	S
Substance	Concentration	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
OTHER ORGANIC SUBSTANCES					,										
Acetone		••	15	• •	15	•••		•••		•••		••		•••	
Ethyl Acetate		•••	15	•••	15	•••		•••		•••		•		•••	
MEK (Methyl Ethyl Ketone)		•••		•••		•••		•••		•••		• •		•••	
Chloroform		•		•		•		•		• •		•		•••	
Cyclohexanone		•••		•••		•••		•••				•••		•••	
Formaldehyde		•••		•••		• •		• •				•••		•••	
Diethyl ether		•••		•••		•••		•••		•••		•••			
Perchloroethylene		••		• •								•		•••	
MISCELLANEOUS															
Fruit juices	Concentrated	•••		•••		•••		•••		•••		•••		•••	
Sanitizers/Biocides for plumbing	Diluted	••	16	• •	16	• •	16	• •	16					•••	16
Heating system fluids (water+glycols)	20%	• •		• •		• •		• •		•••		•••		•••	
Soap (aqueous solution)		•••		•••		•••		•••		•••		•••		•••	
50% Oleic acid + 50% Olive Oil (23°C)		•••		•••		•••		•••		•••				•••	



		radilon A torzen		radilon* S		radilon [*] D		radilon [*] DT		radilon* Aestus		raditer B		radite	
Substance	Concentration	Rank			Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes	Rank	Notes
OTHER		'	'		'	'			'				'		'
Deionized water (90°C)		•••		•••		•••		•••		•••				•••	
Phenol solution (5% by mass) (23°C)		•		•		•		•		•		•		•••	

Legend of Specifications (Notes)

01 = at low conc. <10% may be suitable for limited exposure.

02 = only for short term exposure (<30 days) at room T at conc. below 5%.

03 = only for short term exposure (<60 days) at room T at conc. below 10%.

04 = poor resistance; loss of mechanical properties; swelling.

05 = only for short term exposure (<5 days) at room T then it gets worse.

06 = at very low concentration a proper stabilization can improve the resistance.

07 = subject to environmental stress cracking when under load.

08 = for short to mid term exposure (<300 days).

09 = at room T and not for prolonged contact.

10 = limited change in properties.

11 = for continuous contact with diluted urea solution special grades are available.

12 = glysantin/water 1/1 a 106°C (special grades are available).

13 = swelling is possible.

14 = polymer dissolves at 170°C.

15 = avoid prolonged contact.

16 = check carefully the actual content of the specific product.

17 = diluted (20%).



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